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Key indicators

Single-crystal X-ray study T = 298 K Mean σ (C–C) = 0.006 Å R factor = 0.062 wR factor = 0.210 Data-to-parameter ratio = 14.7

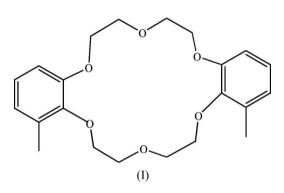
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

cis-3,3'-Dimethyldibenzo-18-crown-6

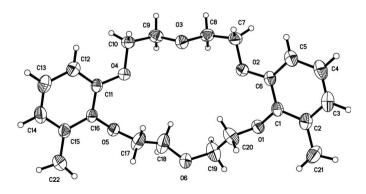
In the title molecule (systematic name: 10,26-dimethyl-2,5,8,15,18,21-hexaoxatricyclo[$20.4.0.0^{9,14}$]tetracosa-9,11,13,-22,24,26-hexaene), C₂₂H₂₈O₆, the two benzene rings make a dihedral angle of 63.9 (2)°, indicating a boat-like molecular conformation. The crystal packing is mainly stabilized by van der Waals forces.

Comment

In continuation of our ongoing programme directed to the development of crown ether chemistry (Niu *et al.*, 2005), we present here the synthesis and crystal structure of the title compound, (I) (Fig. 1).



In (I), the average Csp^2 —O bond length is 1.372 (4) Å (see also Table 1). The interatomic distances O1···O2 [2.664 (4) Å] and O4···O5 [2.641 (4) Å] are somewhat shorter than those reported earlier (Yang *et al.*, 1988). The other intramolecular O···O distances lie in the range 2.941 (4)–3.592 (4) Å. The two benzene rings make a dihedral angle of 63.9 (2)°, indicating a boat-like conformation of the molecule. The crystal packing is mainly stabilized by van der Waals forces.



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Figure 1 The molecular structure of the title compound, showing 30% probability displacement ellipsoids and the atom-numbering scheme.

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Experimental

The reaction was carried out under a nitrogen atmosphere. 3-Methylcatechol (19.84 g, 0.16 mol), sodium hydroxide (6.4 g, 0.16 mol) and dichloroethyl ether (12 g, 0.084 mol) were in turn added to a solution of isopentyl alcohol (80 ml) in a four-mouth flask and the mixture was stirred for 2.5 h. After cooling to room temperature, the mixture was filtered. The solid was recrystallized from *n*-heptane to obtain crystals of (I) suitable for X-ray diffraction.

Crystal data

$C_{22}H_{28}O_{6}$	Z = 4
$M_r = 388.44$	$D_x = 1.215 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
a = 9.354 (3) Å	$\mu = 0.09 \text{ mm}^{-1}$
b = 7.828 (3) Å	T = 298 (2) K
c = 29.122 (11) Å	Block, colourless
$\beta = 95.386 \ (5)^{\circ}$	$0.40 \times 0.32 \times 0.28 \text{ mm}$
V = 2123.0 (13) Å ³	

Data collection

Bruker SMART CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\min} = 0.966, T_{\max} = 0.976$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.063$ $wR(F^2) = 0.210$ S = 1.003714 reflections 253 parameters H-atom parameters constrained $0.40 \times 0.32 \times 0.28$ mm 10489 measured reflections 3714 independent reflections

1897 reflections with $I > 2\sigma(I)$

 $R_{\rm int} = 0.042$

 $\theta_{\rm max} = 25.0^{\circ}$

$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0806P)^{2} + 1.8496P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3 (\Delta/\sigma)_{max} < 0.001 \Delta\rho_{max} = 0.42 \text{ e} \text{ Å}^{-3}$

 $\Delta \rho_{\rm min} = -0.25 \text{ e} \text{ Å}^{-3}$

Table 1

Selected bond lengths (Å).

O1-C1	1.375 (4)	O3-C8	1.415 (5)
O1-C20	1.435 (5)	O3-C9	1.416 (5)
O2-C6	1.369 (4)	O4-C11	1.365 (4)
O2-C7	1.424 (4)	O4-C10	1.426 (4)

All H atoms were placed geometrically (C-H = 0.93-0.97 Å) and treated as riding on their parent atoms with $U_{iso}(H) = 1.2U_{eq}(C)$ or $1.5U_{eq}(C)$.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

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References

Niu, M. J., Wang, D. Q., Li, D. C. & Dou, J. M. (2005). Z. Kristallogr. New Cryst. Struct. 220, 188–190.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Software Reference Manual. Bruker AXS Inc., Madison, Wisconsin, USA.

Siemens (1996). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Yang, G. D., Sun, Y. X., Zhang, H. X. & Zhang, H. (1988). Chem. J. Chin. Univ. 9, 618–621.